

# **ET Theory 101**

## **USCID Workshop**

**<http://biomet.ucdavis.edu>**

**PMhr, PMday, PMmon**

**CUP, SIMETAW (DWR link)**

# Methods of Heat Transfer

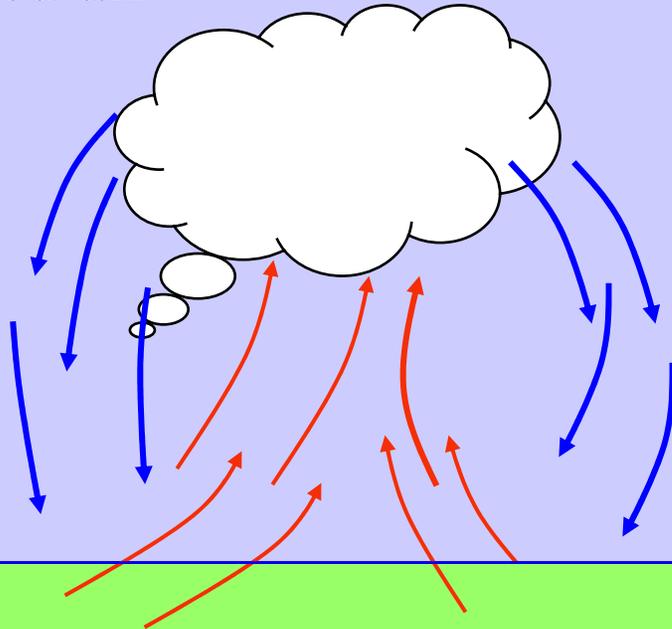
**Conduction**- from molecule to molecule

**Heat Source**

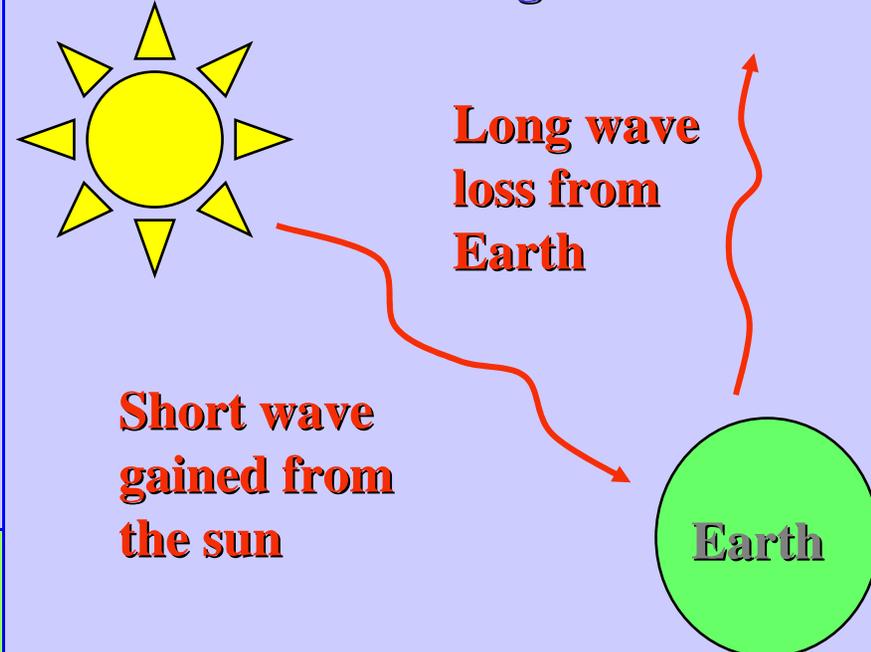


**Metal bar**

**Convection** - by movement of heated air



**Radiation** - energy passing from one object to another without a connecting medium



# Questions 1 and 2

1. How many molecules of air are in a cubic meter?

$$2.69 \times 10^{25}$$

26,900,000,000,000,000,000,000,000 molecules m<sup>-3</sup>

2. The air is mostly nitrogen? (T/F)?

False

1. If six billion people count 1000 molecules per hour, it would take 511 million years to count the number of molecules in 1 m<sup>3</sup>

2. Air is mostly empty space. Less than 0.1% of the volume is occupied by molecules

## *Questions 3 and 4*

**3. Which molecules move faster?**

- A. nitrogen**
- B. oxygen**
- C. carbon dioxide**
- D. water vapor**

**4. When saturated, the air cannot hold more water vapor (T/F)?**

**3. At 30°C, velocities**

**N<sub>2</sub> - 1709 Km h<sup>-1</sup>**

**O<sub>2</sub> - 1603 Km h<sup>-1</sup>**

**CO<sub>2</sub> - 1366 Km h<sup>-1</sup>**

**H<sub>2</sub>O - 2136 Km h<sup>-1</sup>**

**4. False! Less than 0.1% of volume is occupied. Air can hold more water.**

# Questions 5

5. How many molecule collisions per second occur in a  $\text{m}^3$  of air?

5. There are about 595 trillion collisions per second between air molecules in a  $\text{m}^3$  of air.

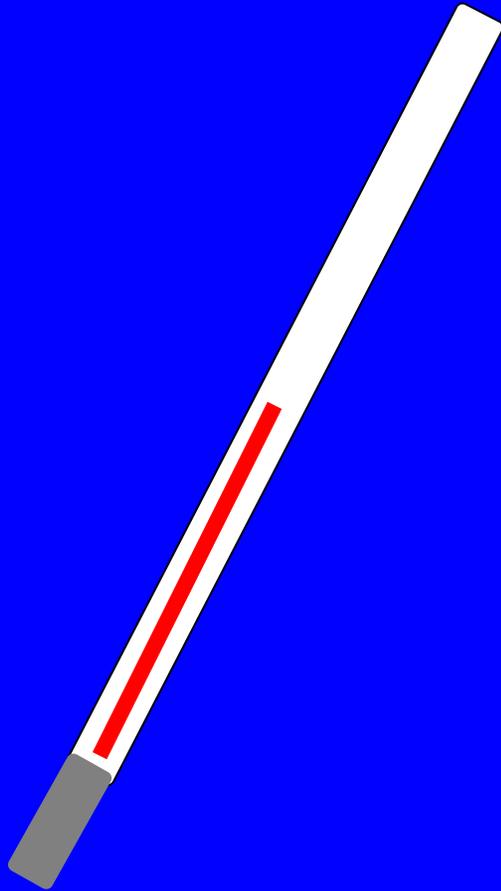
26,900,000,000,000,000,000,000,000 molecules  $\text{m}^{-3}$

Occupy < 0.1% volume

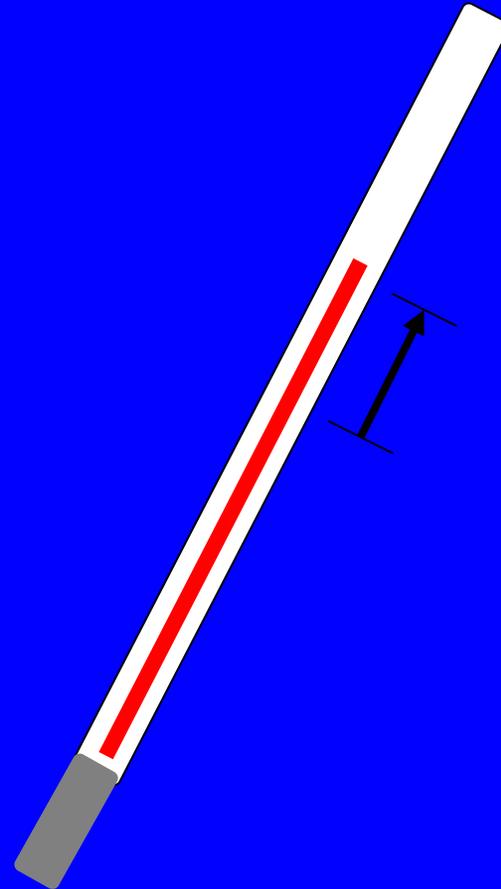
595,000,000,000,000 collisions/sec

# Sensible Heat & Temperature

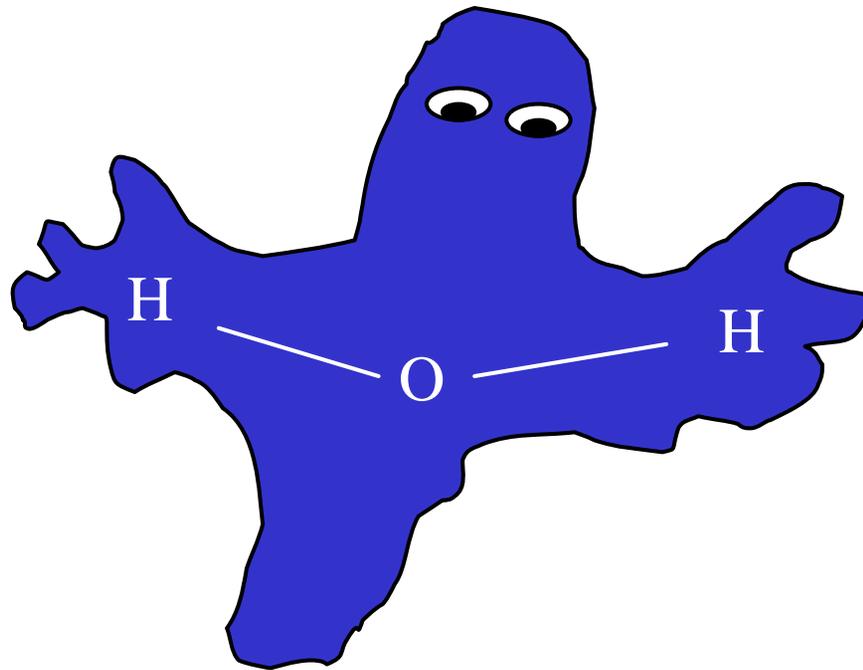
Lower temperature  
Less **sensible heat**



Higher temperature  
More **sensible heat**



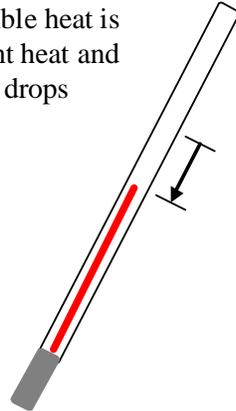
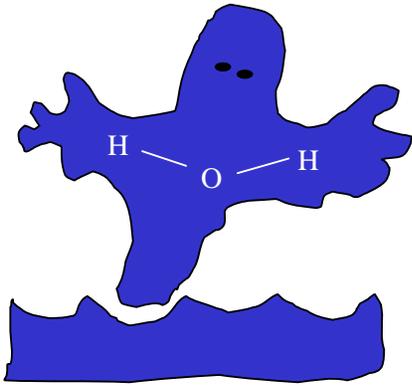
Water Molecules consist  
of one oxygen and two  
hydrogen atoms



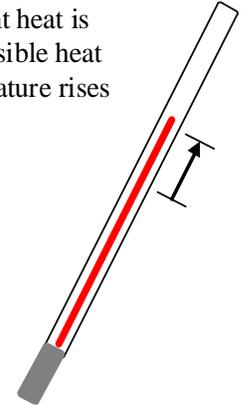
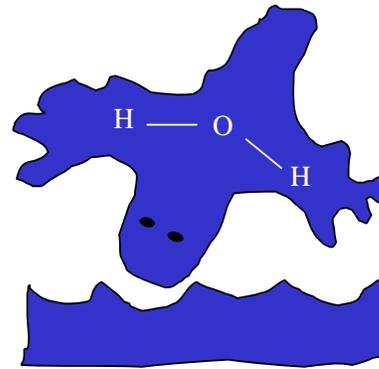
# Methods of Heat Transfer

## Latent Heat - Chemical Heat

When water molecules evaporate, sensible heat is changed to latent heat and the temperature drops



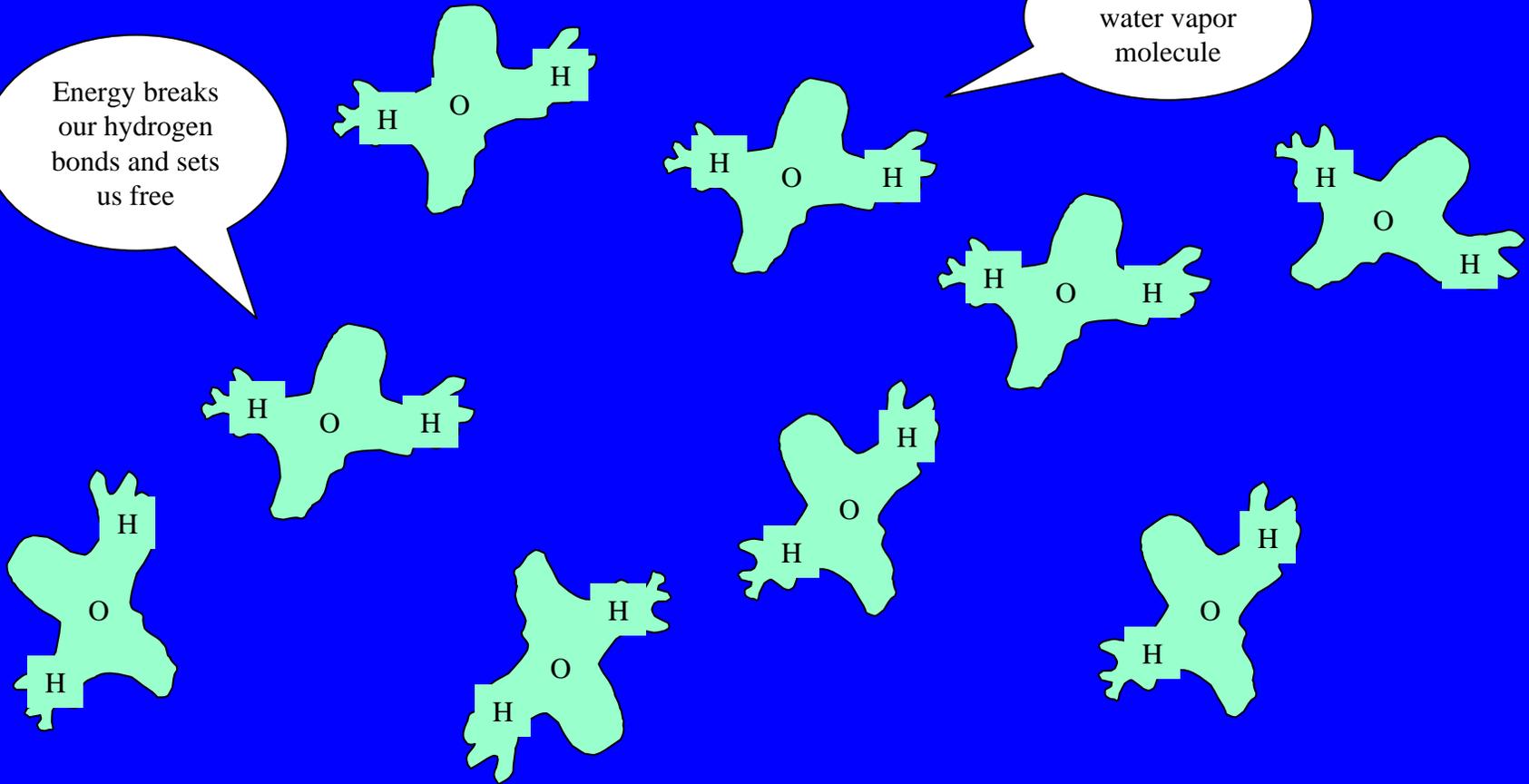
When water molecules condense, latent heat is changed to sensible heat and the temperature rises



# More Water Vapor More Latent Heat

Energy breaks our hydrogen bonds and sets us free

I am a free water vapor molecule



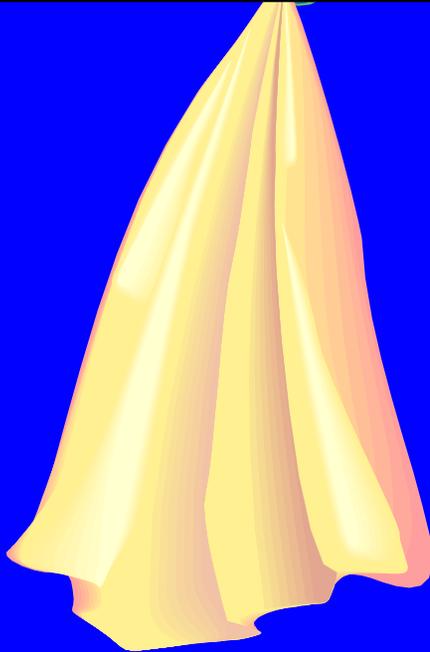
Latent heat

## *Questions 6*

**6. Your ice cream will melt faster at 35% RH than at 75% RH (T/F)?**

**6. False, there is more total heat in humid than in dry air to melt the ice cream**

# Which wet towel dries fastest?



30 % RH



70 % RH

$T_a = 30\text{ }^\circ\text{C}$

Adiabatic Process

# *Questions 7*

**7. Is there evaporation in a greenhouse with 100% relative humidity?**

**7. Yes! But both the temperature and vapor pressure must increase following the saturation vapor pressure curve.**

## **Diabatic Process**

# Radiation

$$E = \varepsilon \sigma T^4$$

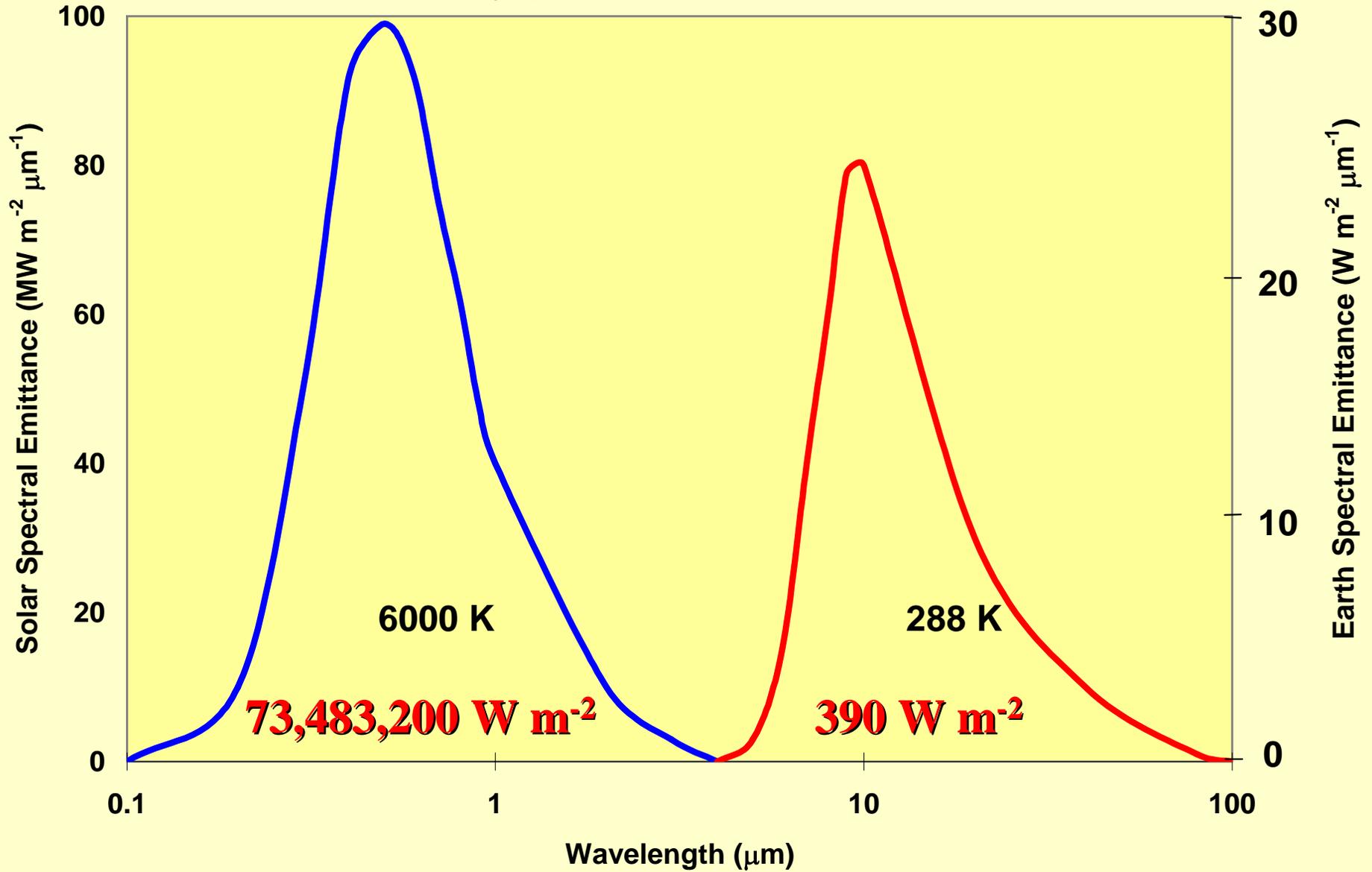
$\varepsilon = 1.0$  for a black body

$\varepsilon < 1.0$  for a gray body

$$\sigma = 5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-1}$$

$T = \text{absolute temperature} = \text{°C} + 273.15$

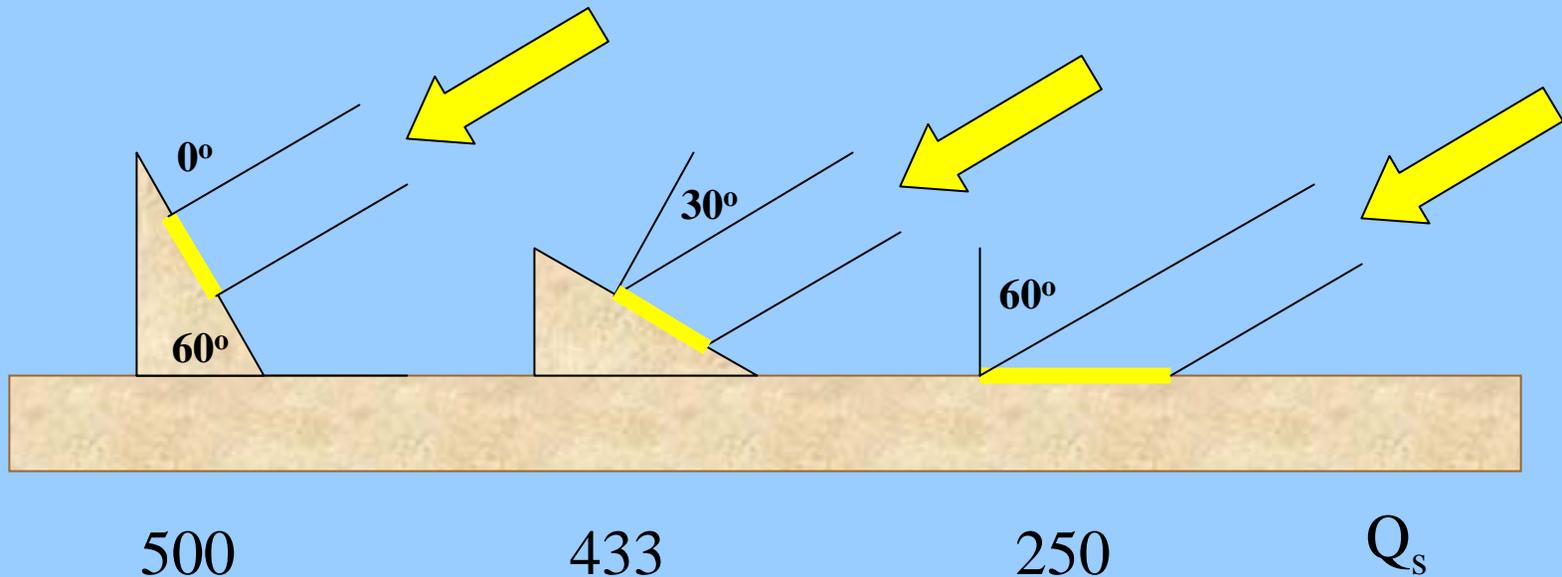
Blackbody spectral emittance for Sun and Earth



# Direct Radiation ( $Q$ )

- Radiation that comes directly from the sun
- The amount of energy received per unit area is called “Irradiance” and the units are commonly  $W\ m^{-2} = J\ s^{-1}\ m^{-2}$ .
- The amount of energy received depends on the angle of incidence of the radiation

# Direct beam radiation interception ( $\text{W m}^{-2}$ )

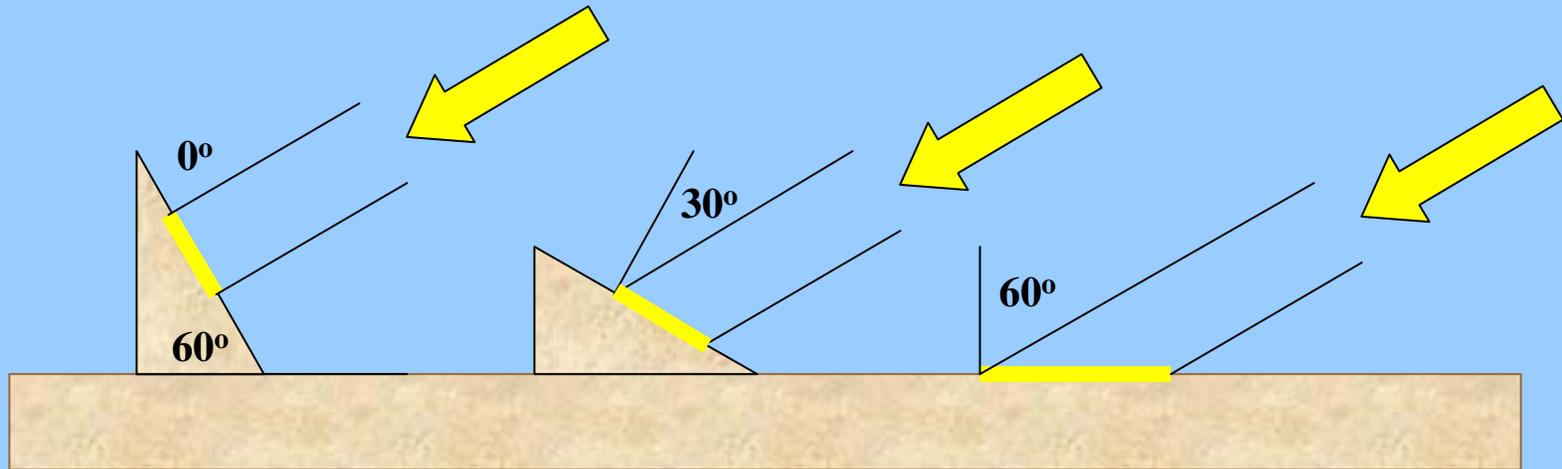


$$Q_s = Q \cos(\alpha)$$

# Diffuse Radiation ( $q$ )

- **Solar radiation that is scattered by the sky and comes from all directions.**
- **During much of the day, about 10 to 15% of the radiation received on a flat surface is diffuse radiation.**
- **The diffuse radiation is nearly the same regardless of the surface orientation.**
- **The percentage of diffuse radiation is higher near sunrise and sunset.**

# Direct and diffuse radiation ( $\text{W m}^{-2}$ )



500

433

250

$Q_s$

535

468

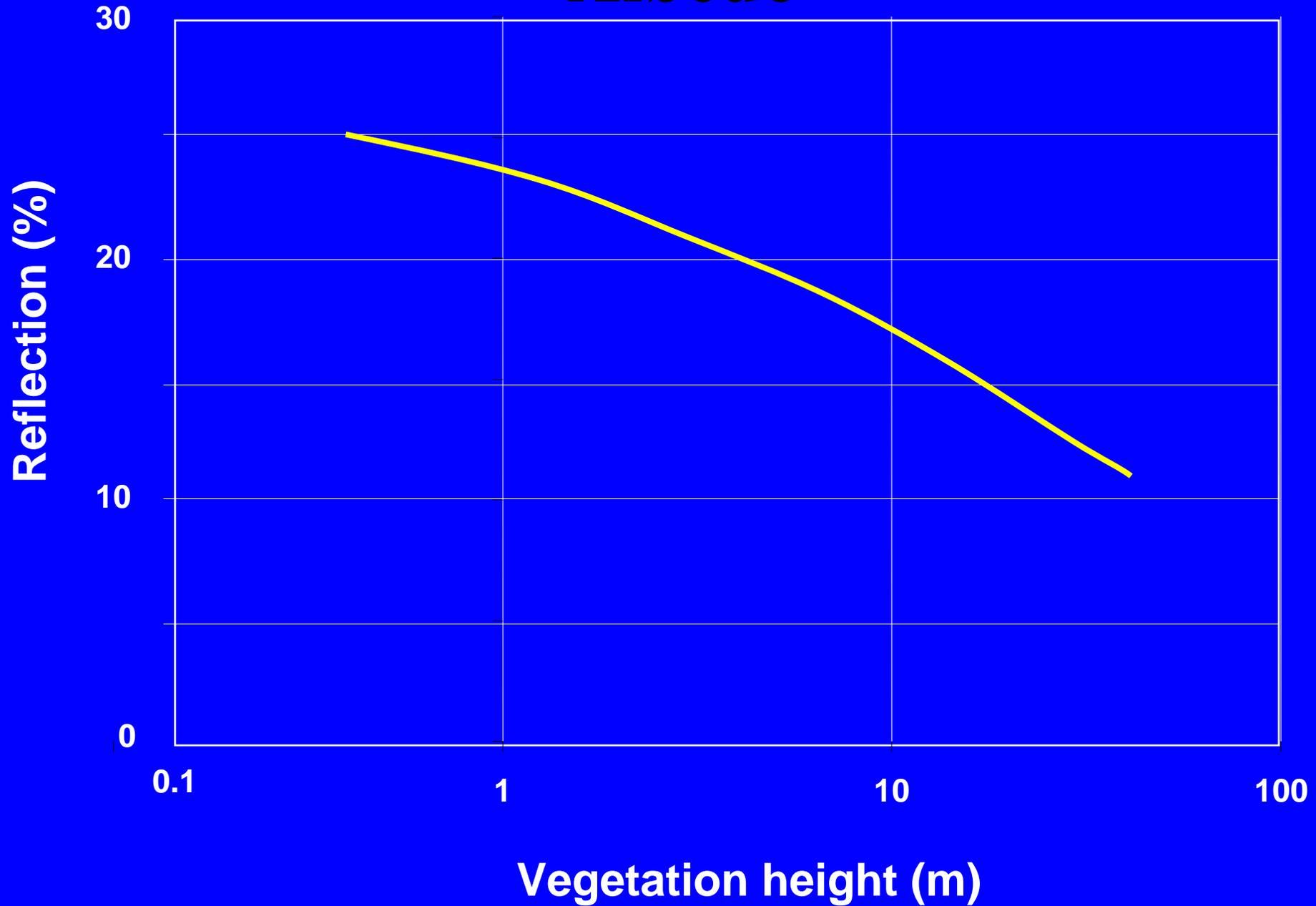
285

$Q_s + q$

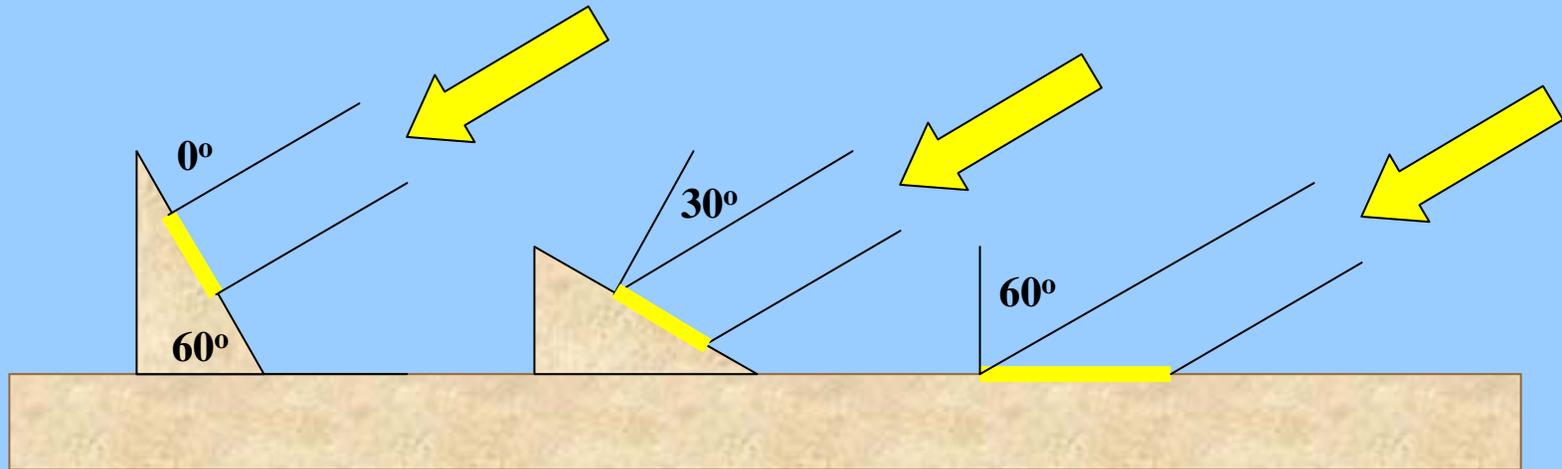
$$R_s = Q_s + q$$

$q \approx 0.15 Q_s$  on a horizontal sfc

# Albedo



# Net solar radiation ( $\text{W m}^{-2}$ )



500

433

250

$Q_s$

535

468

285

$Q_s + q$

401

351

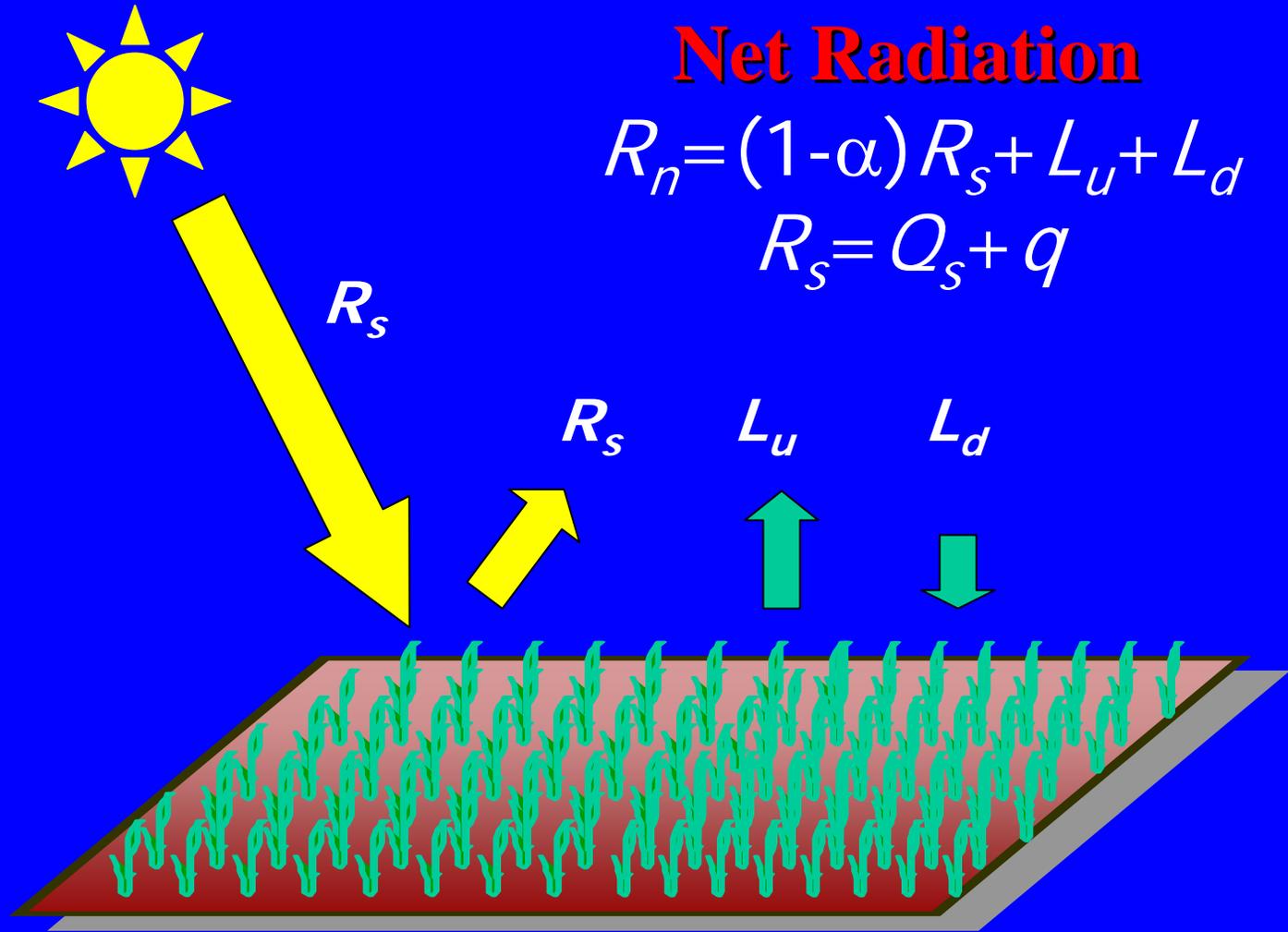
214

$\alpha = 0.25$

# Net Radiation

$$R_n = (1 - \alpha) R_s + L_u + L_d$$

$$R_s = Q_s + q$$



*cloudy*

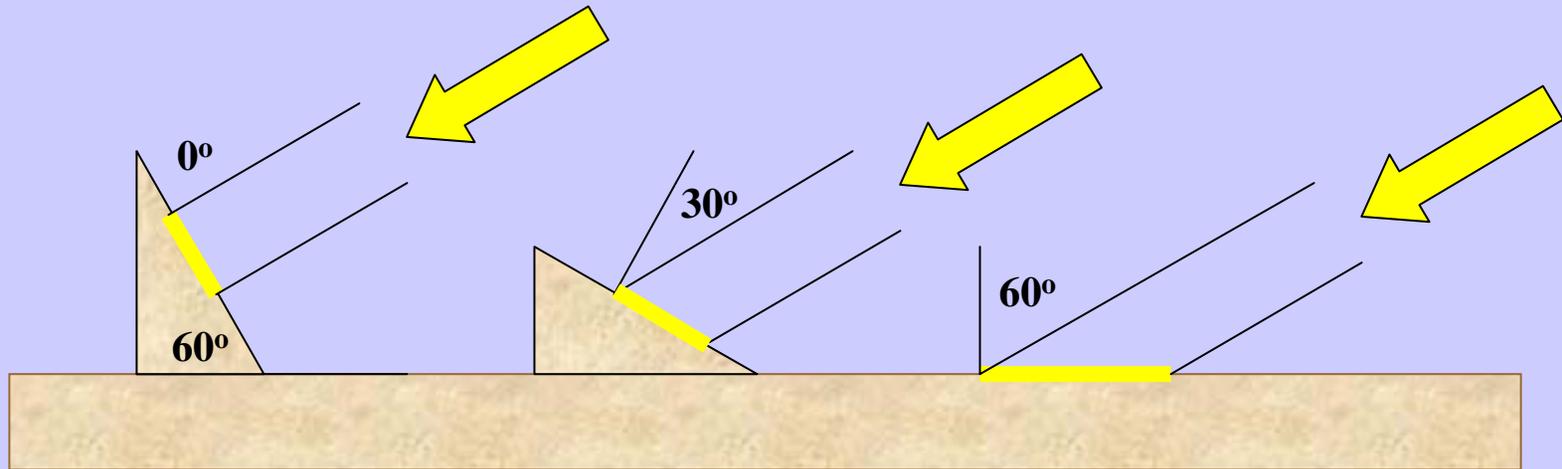
*clear*

$$L_u + L_d = -10 \text{ W m}^{-2}$$

to

$$L_u + L_d = -100 \text{ W m}^{-2}$$

# Net radiation ( $\text{W m}^{-2}$ )



500

433

250

$Q_s$

535

468

285

$Q_s + q$

401

351

214

$\alpha = 0.25$

301

251

114

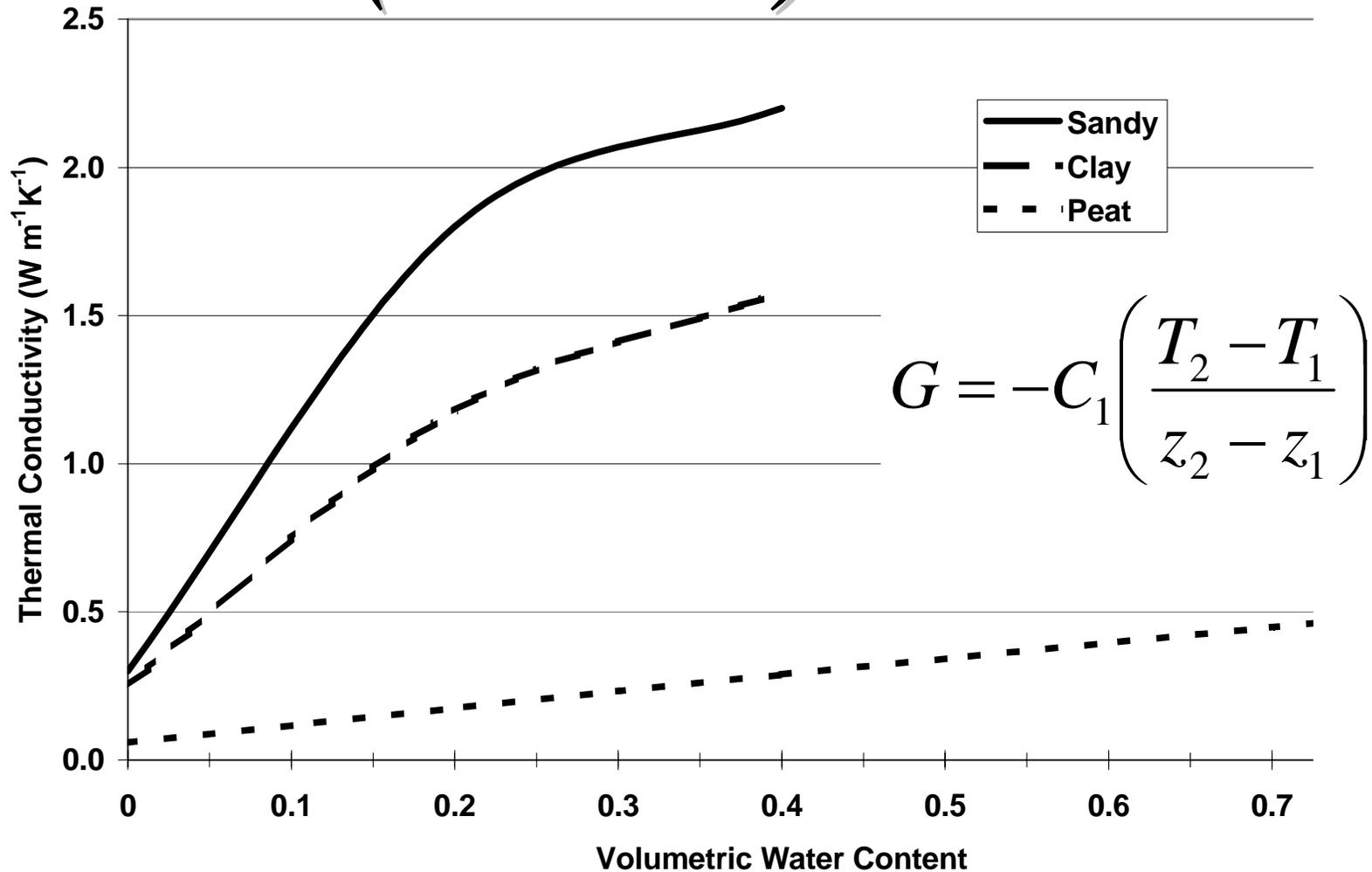
$R_{Ln} = -100$

$$R_n = R_{sd} (1 - \alpha) + R_{Ld} + R_{Lu}$$

# Net Radiometer



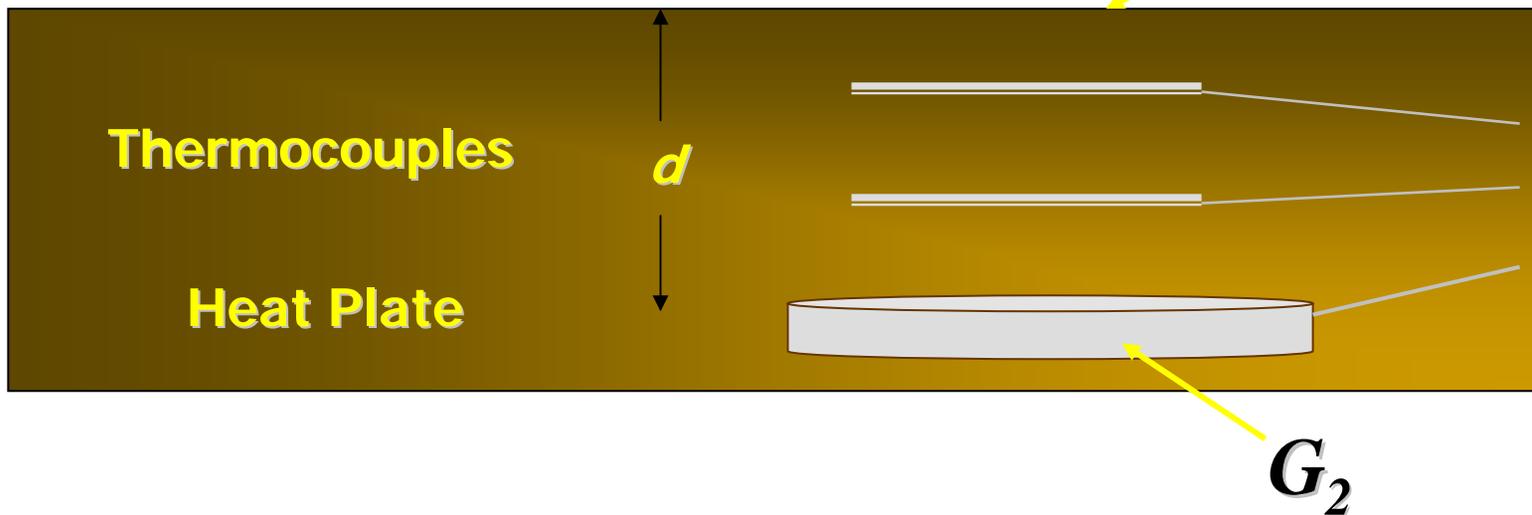
# Soil (Ground) Heat Flux



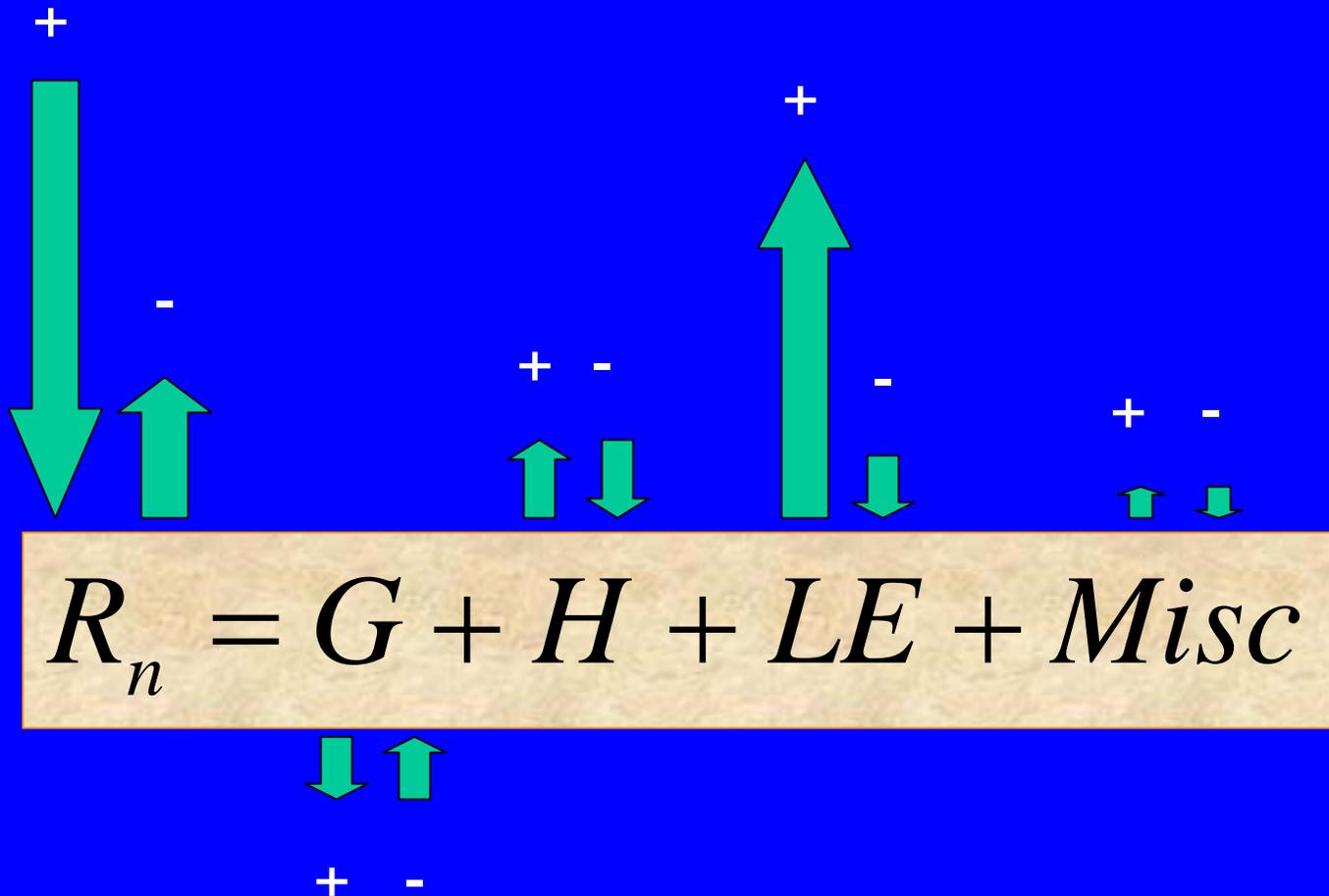
# Soil Heat Flux Density ( $G$ )

$$G = G_2 + C_V \left( \frac{T_f - T_i}{t_f - t_i} \right) d$$

$C_V$  – volumetric heat capacity



# SURFACE ENERGY BUDGETS



# Sensible Heat Flux Density

$$H = -\rho C_p \kappa \left( \frac{T_2 - T_1}{z_2 - z_1} \right)$$

$$H = -\rho C_p \left( \frac{T_2 - T_1}{r_h} \right)$$

$$\frac{1}{r_h} = \frac{\kappa}{z_2 - z_1} = g_h$$

# Latent Heat Flux Density

$$LE = -\kappa \left( \frac{\rho C_p}{\gamma} \right) \left( \frac{e_2 - e_1}{z_2 - z_1} \right)$$

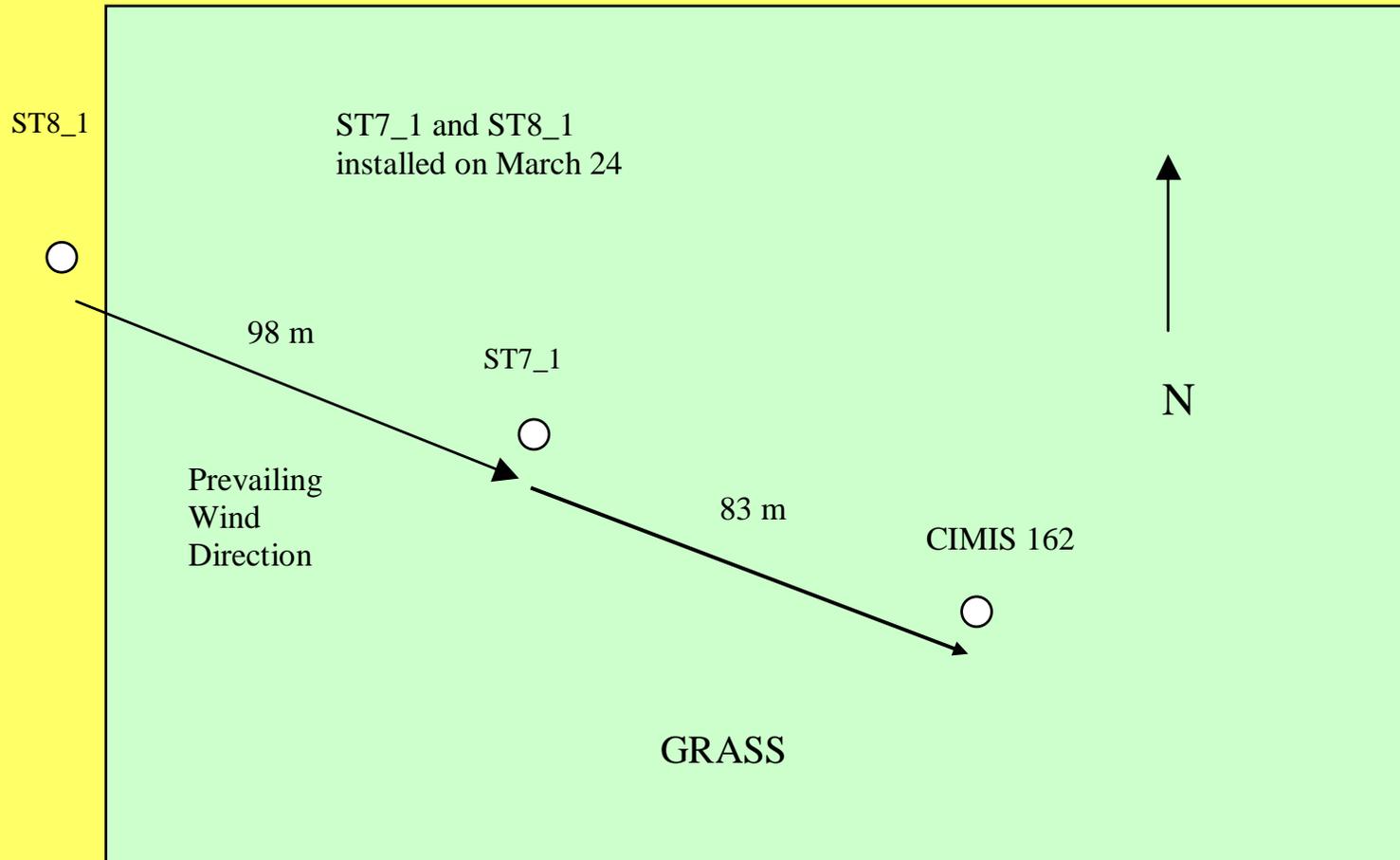
$$LE = - \left( \frac{\rho C_p}{\gamma} \right) \left( \frac{e_2 - e_1}{r_w} \right)$$

$$\frac{1}{r_w} = \frac{\kappa}{z_2 - z_1} = g_w$$

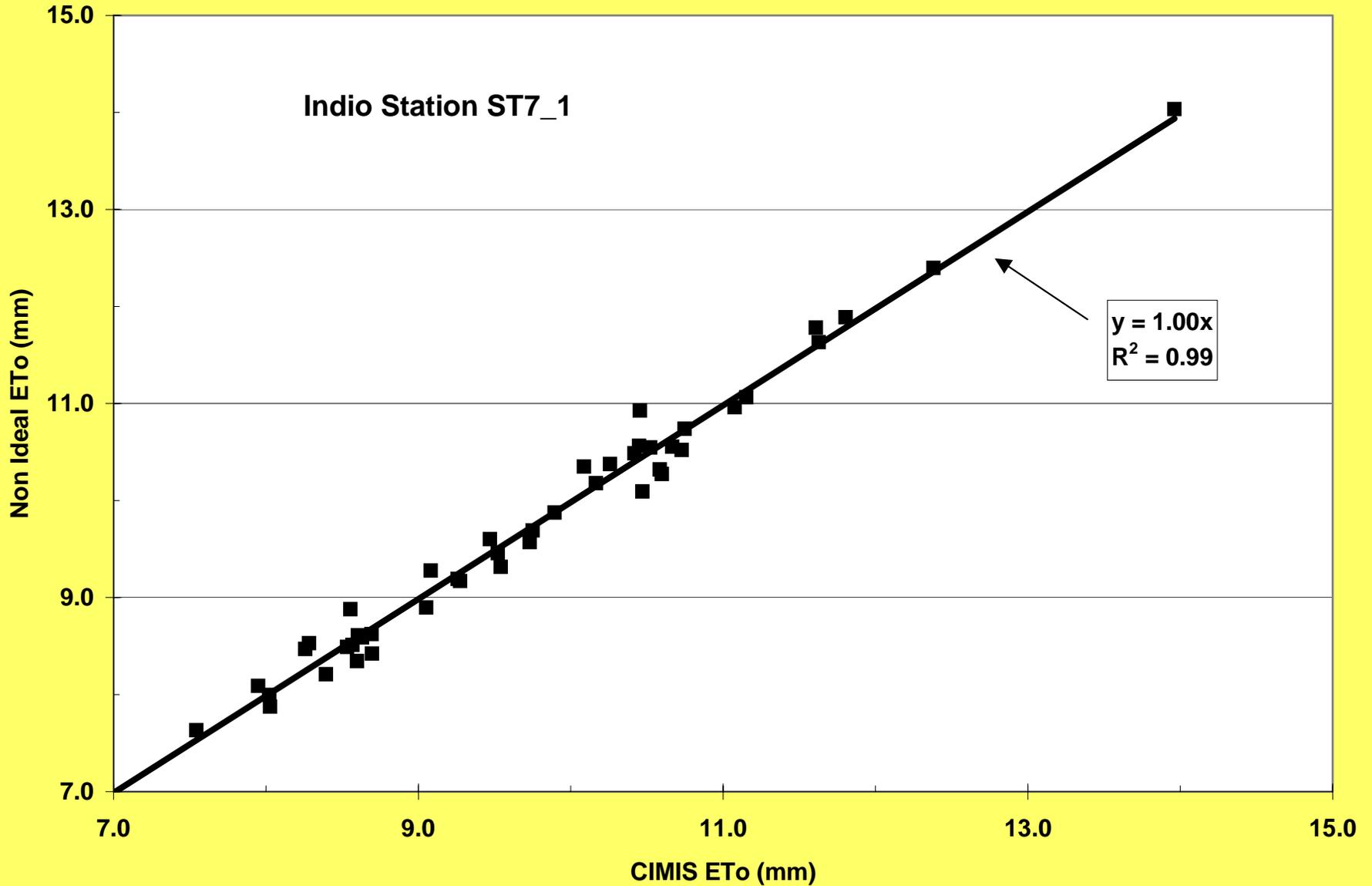
# **Microclimate & ETo**

- **Fetch Requirements**
- **Shading**
- **Wind blocking**
- **Marine effects**

# Fetch Requirements (Indio)



# Fetch (98 m Vs 181 m)



# Empirical ETo – El Dorado Country Club



**Adjustments for  
Regional effects**

**Wind Blockage**

**Sunrise-Sunset**

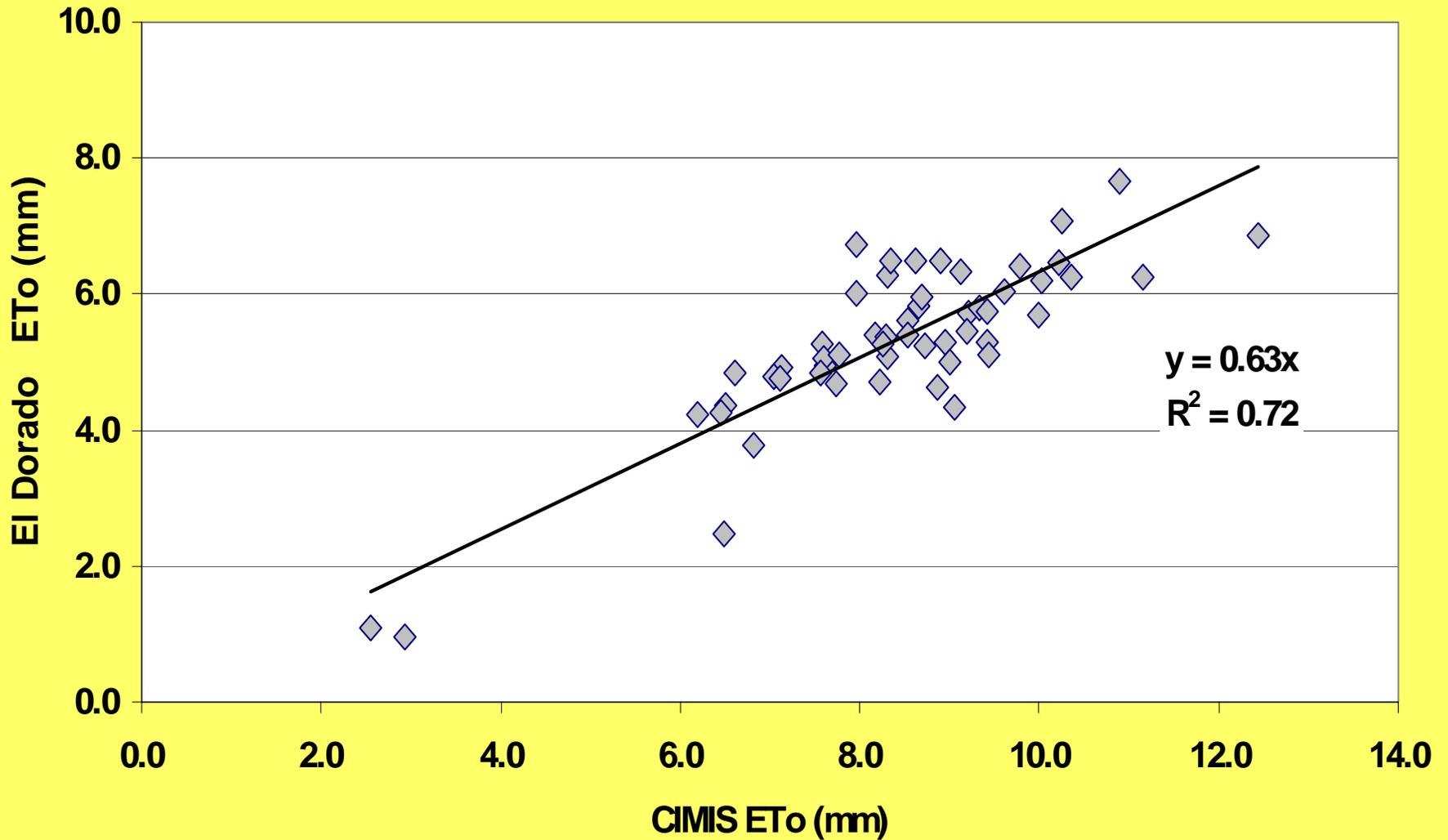
**Advection**

**Empirical  
Equations**

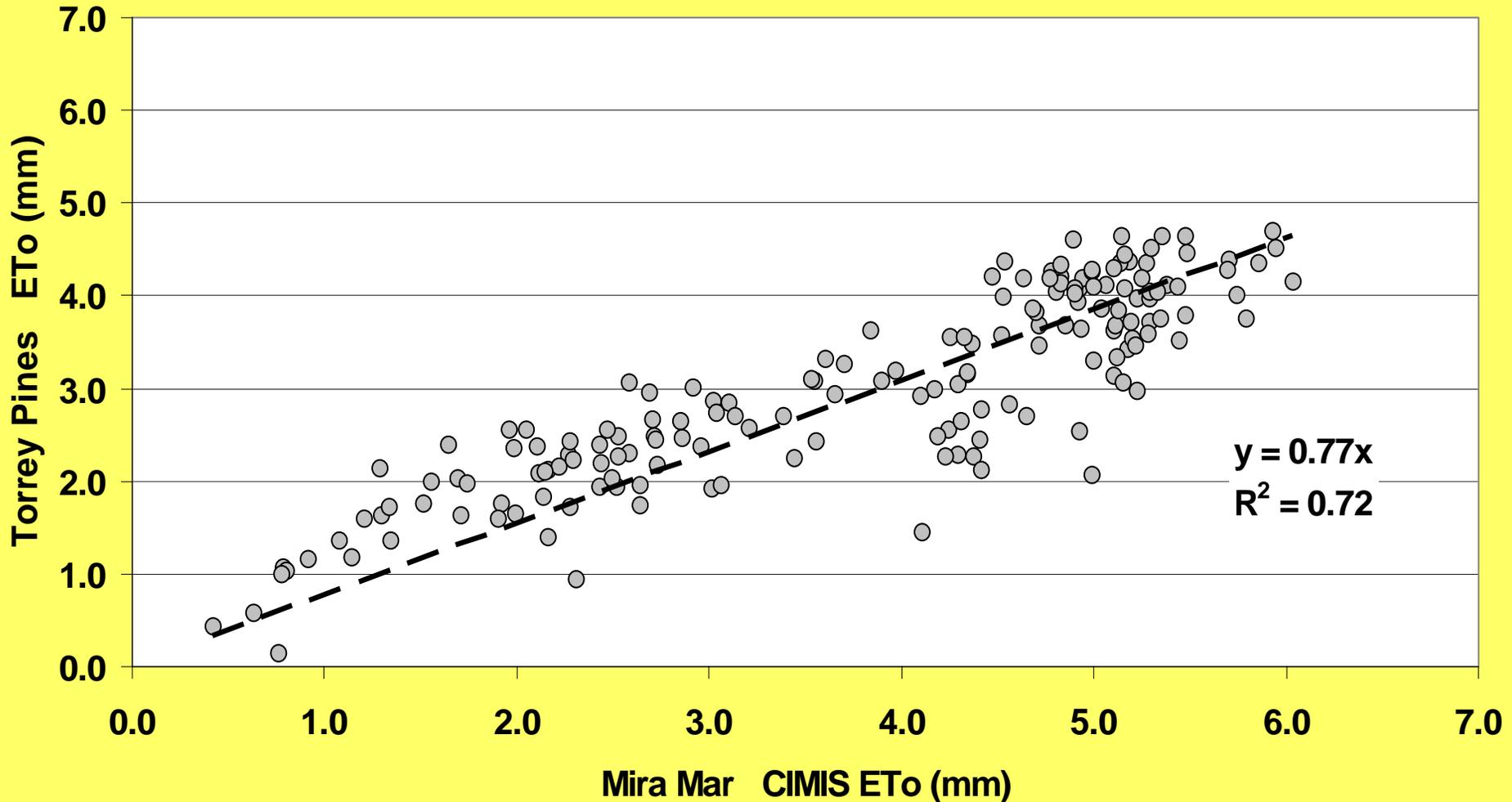
**CIMIS**

**Regional effects**

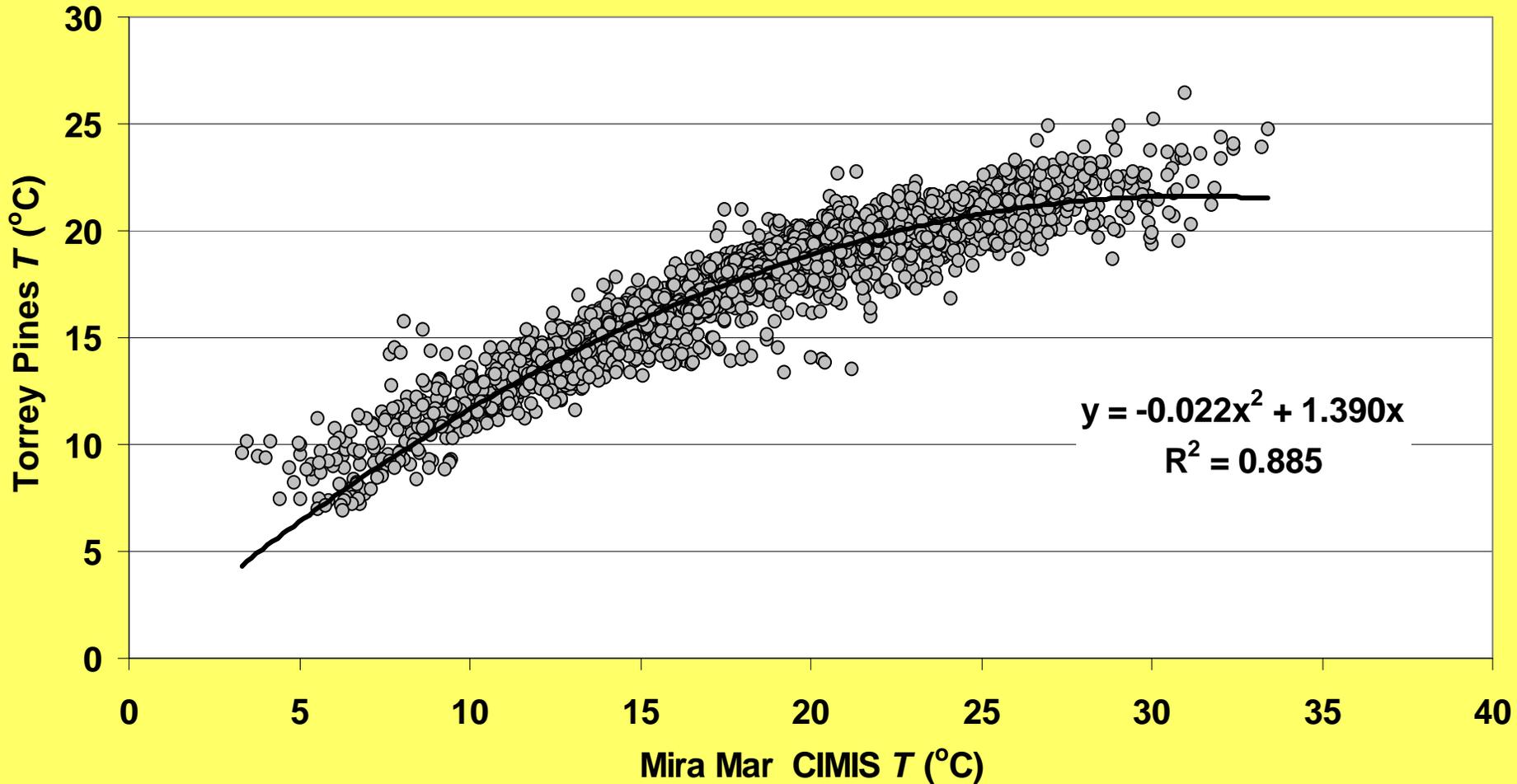
# El Dorado Country Club



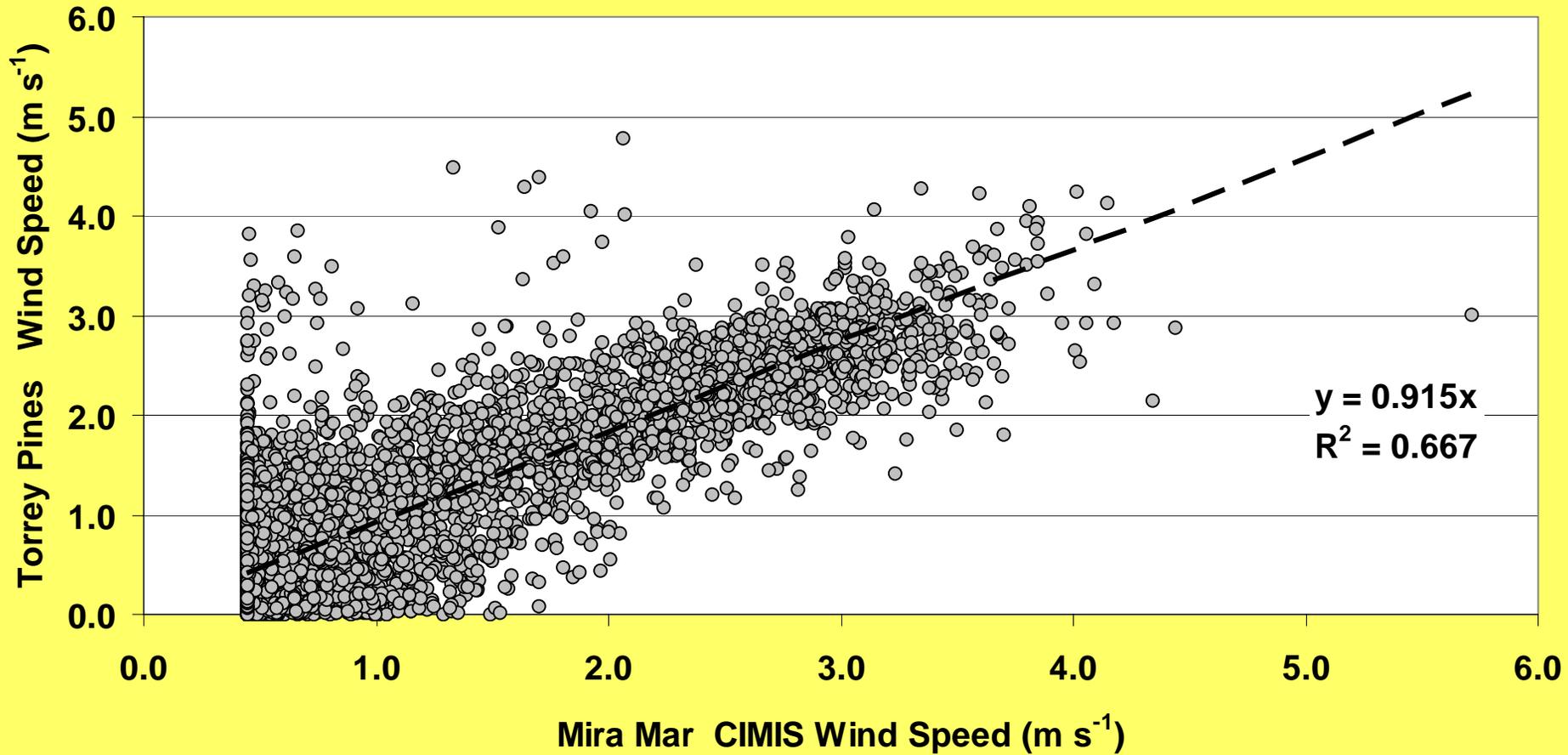
# Torrey Pines Vs Mira Mar



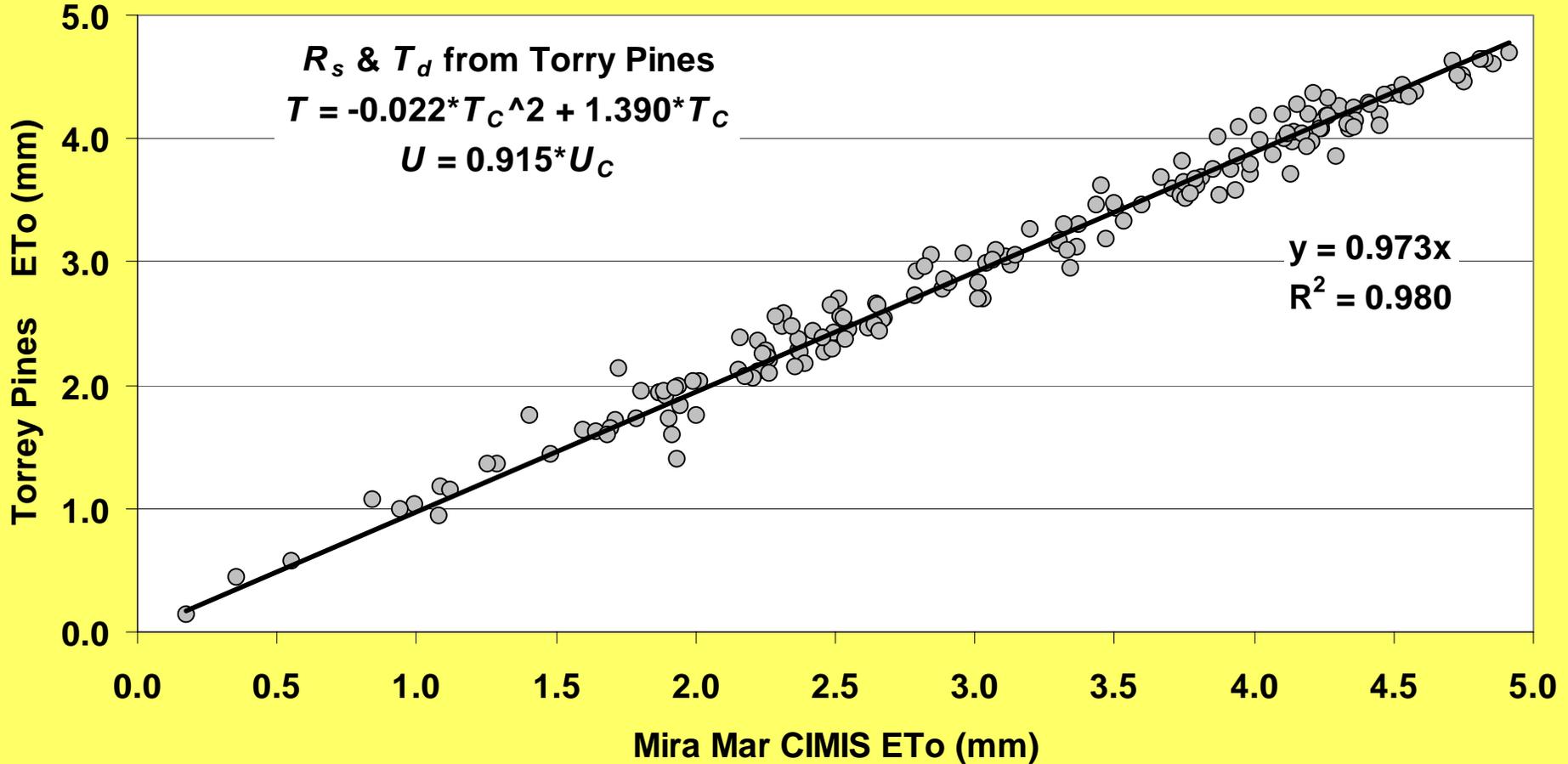
# Temperature Model



# Wind Correction



# Torrey Pines ETo Model

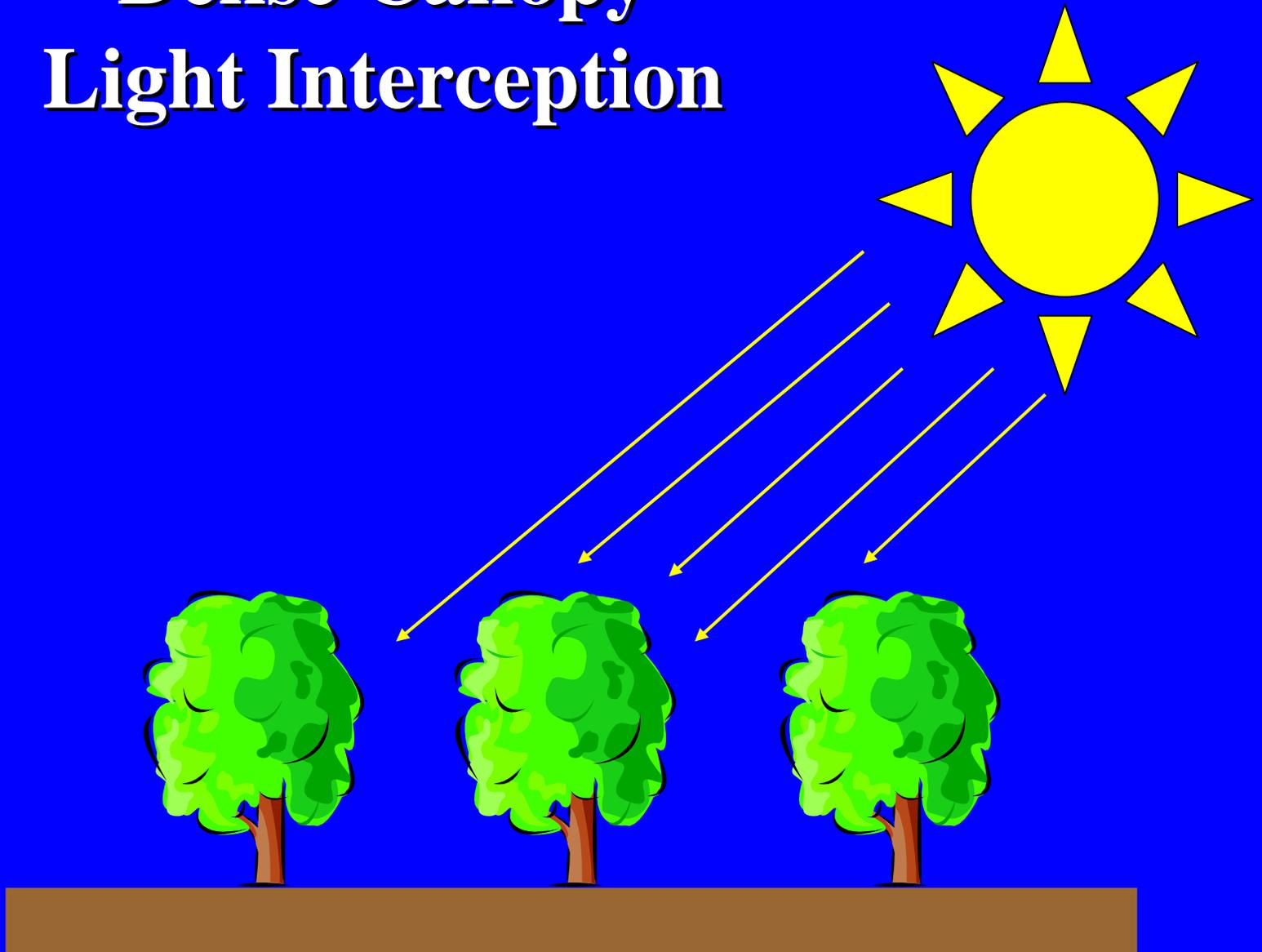


# Landscape Coefficient

$$K_L = \frac{ET_L}{ET_o}$$

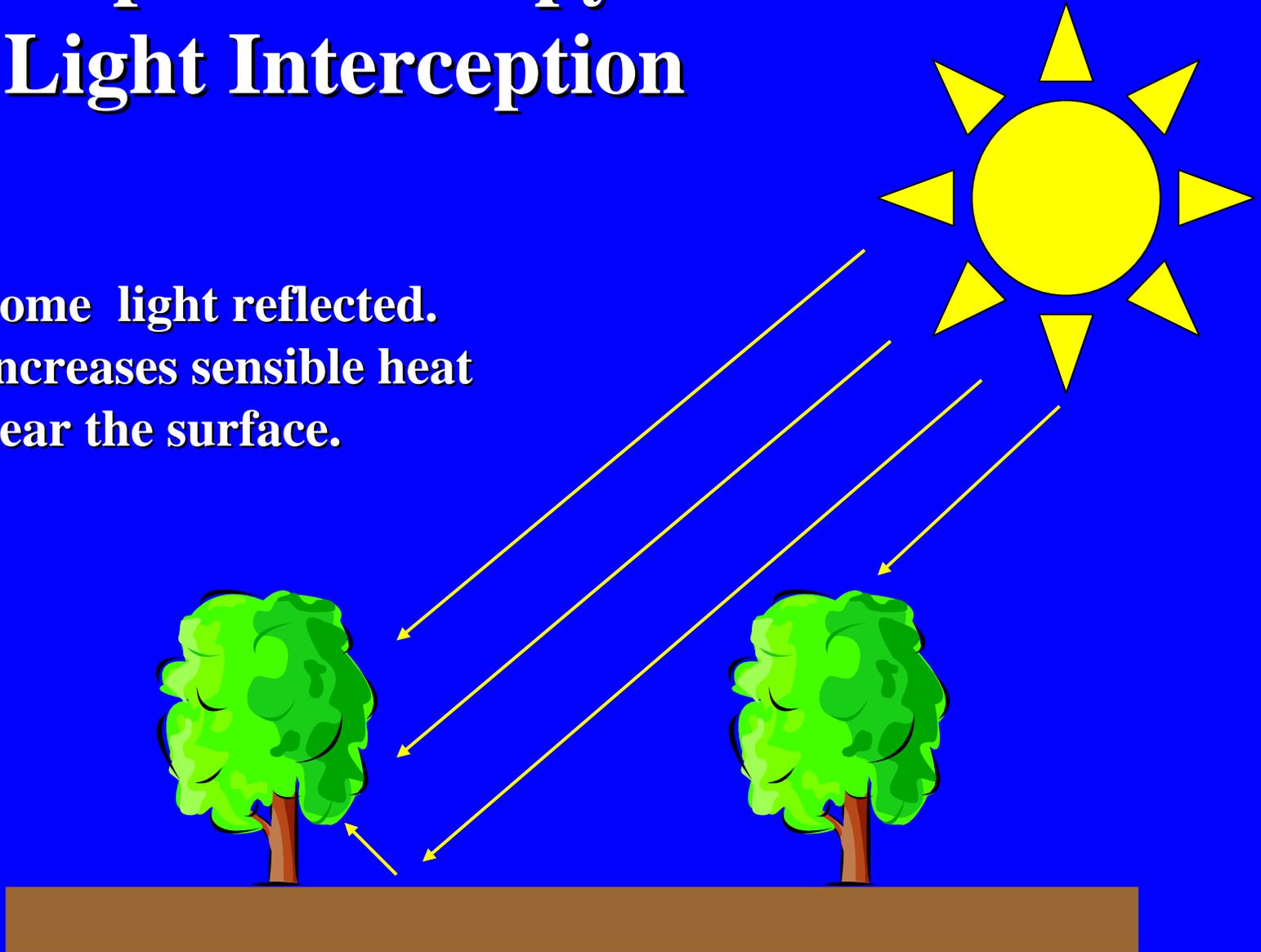
$ET_L$  - measured       $ET_o$  - estimated

# Dense Canopy Light Interception

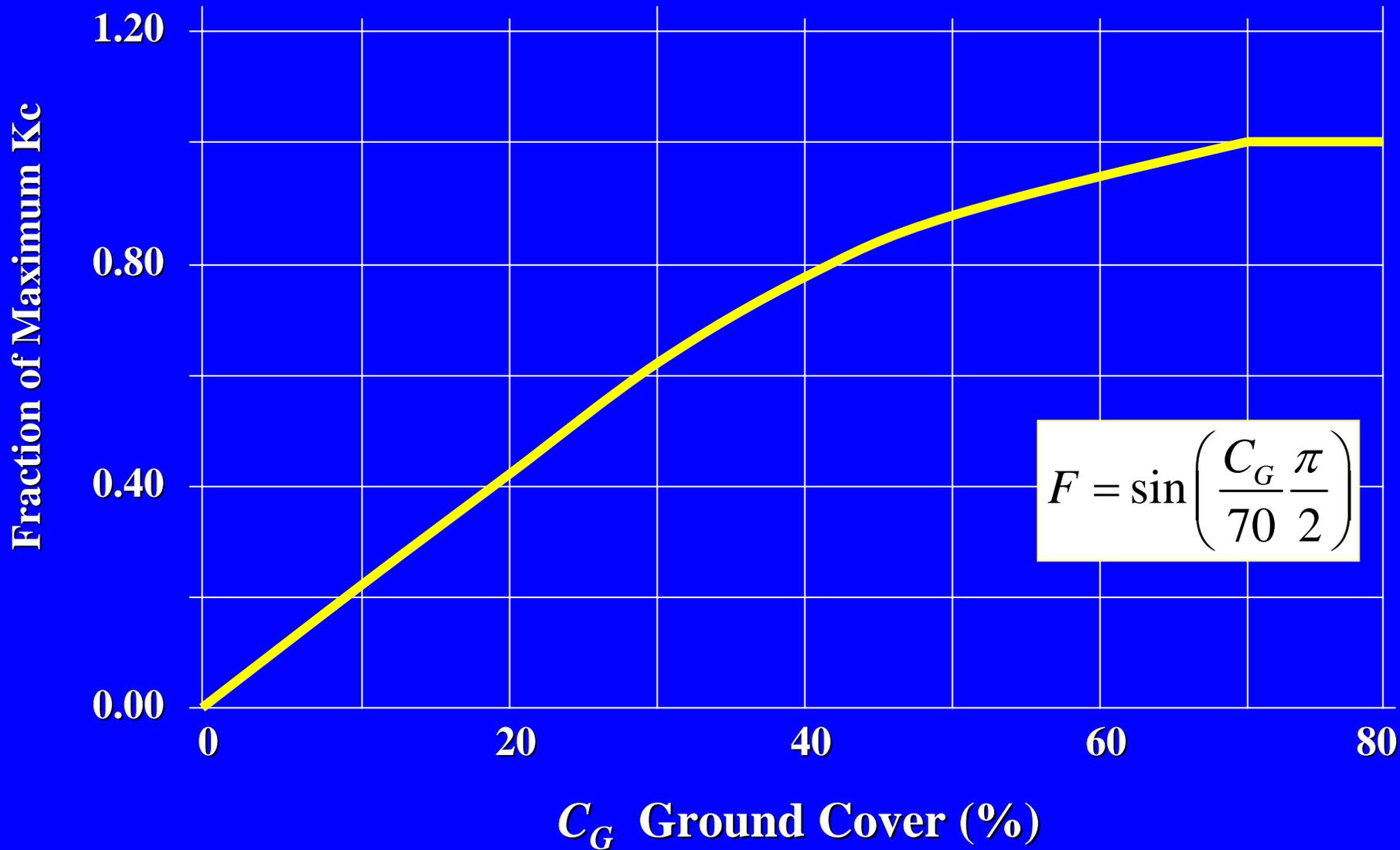


# Sparse Canopy Light Interception

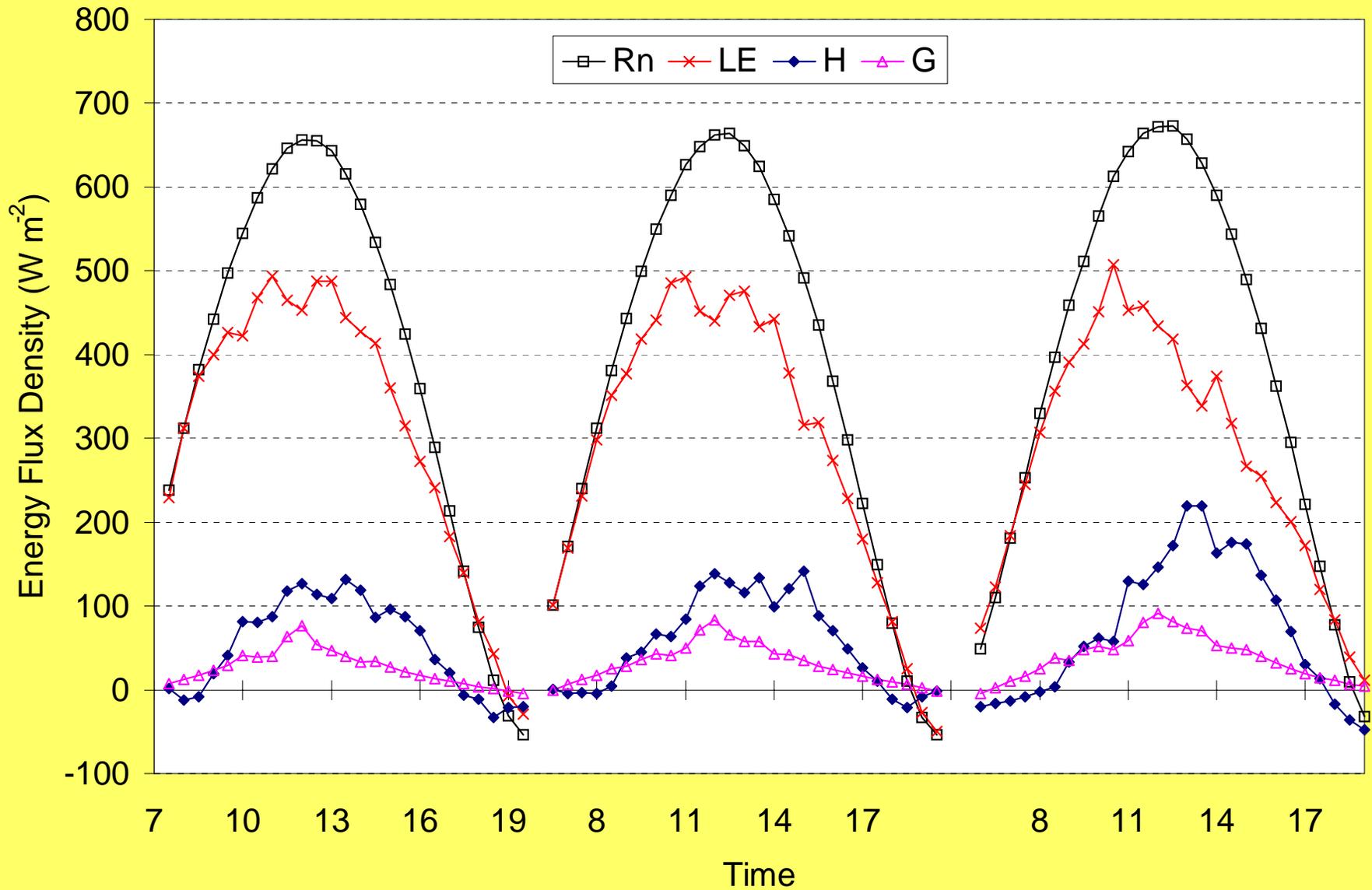
Some light reflected.  
Increases sensible heat  
near the surface.



# Density Correction

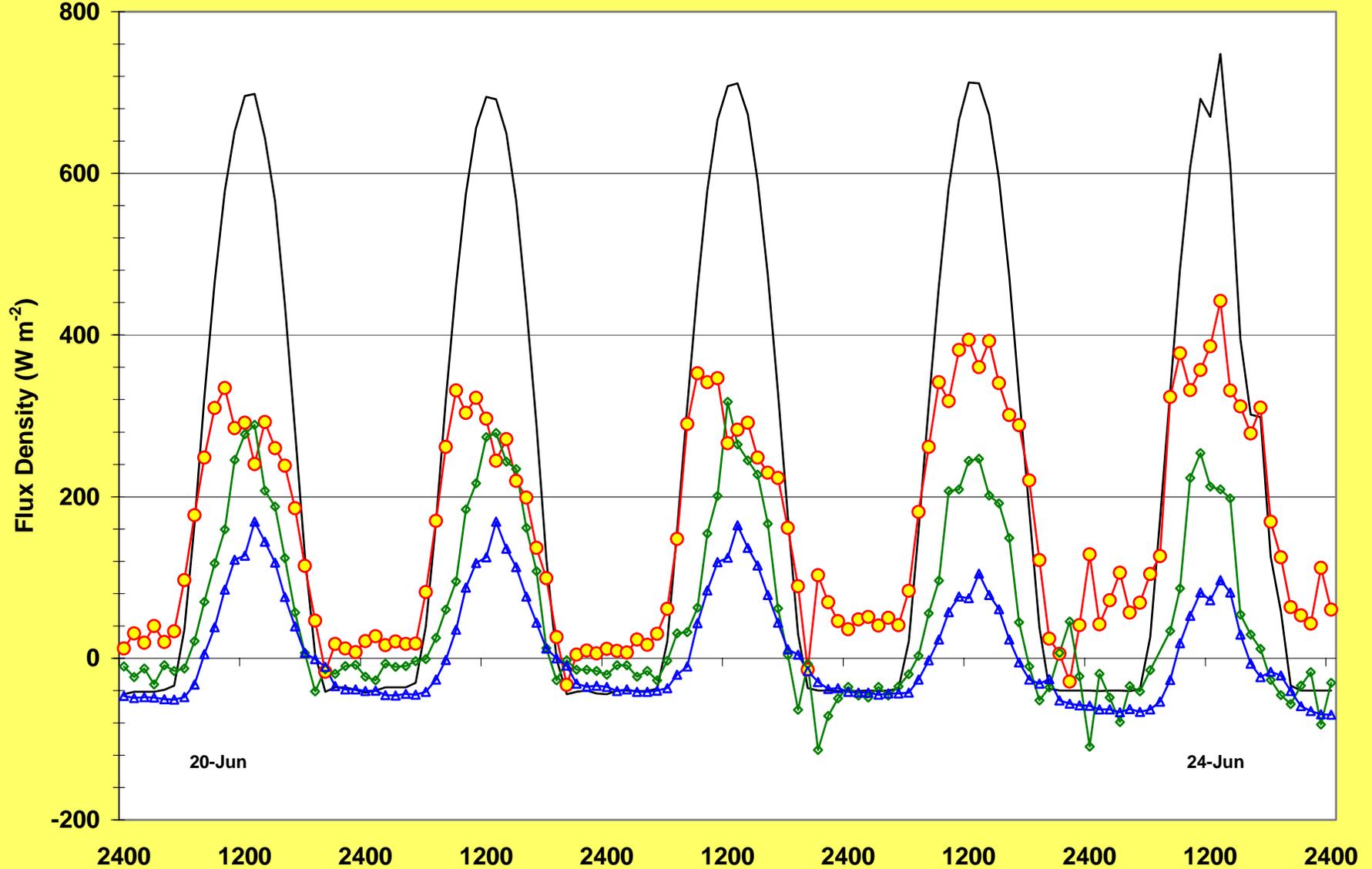


# Sunflowers (Bari, Italy)

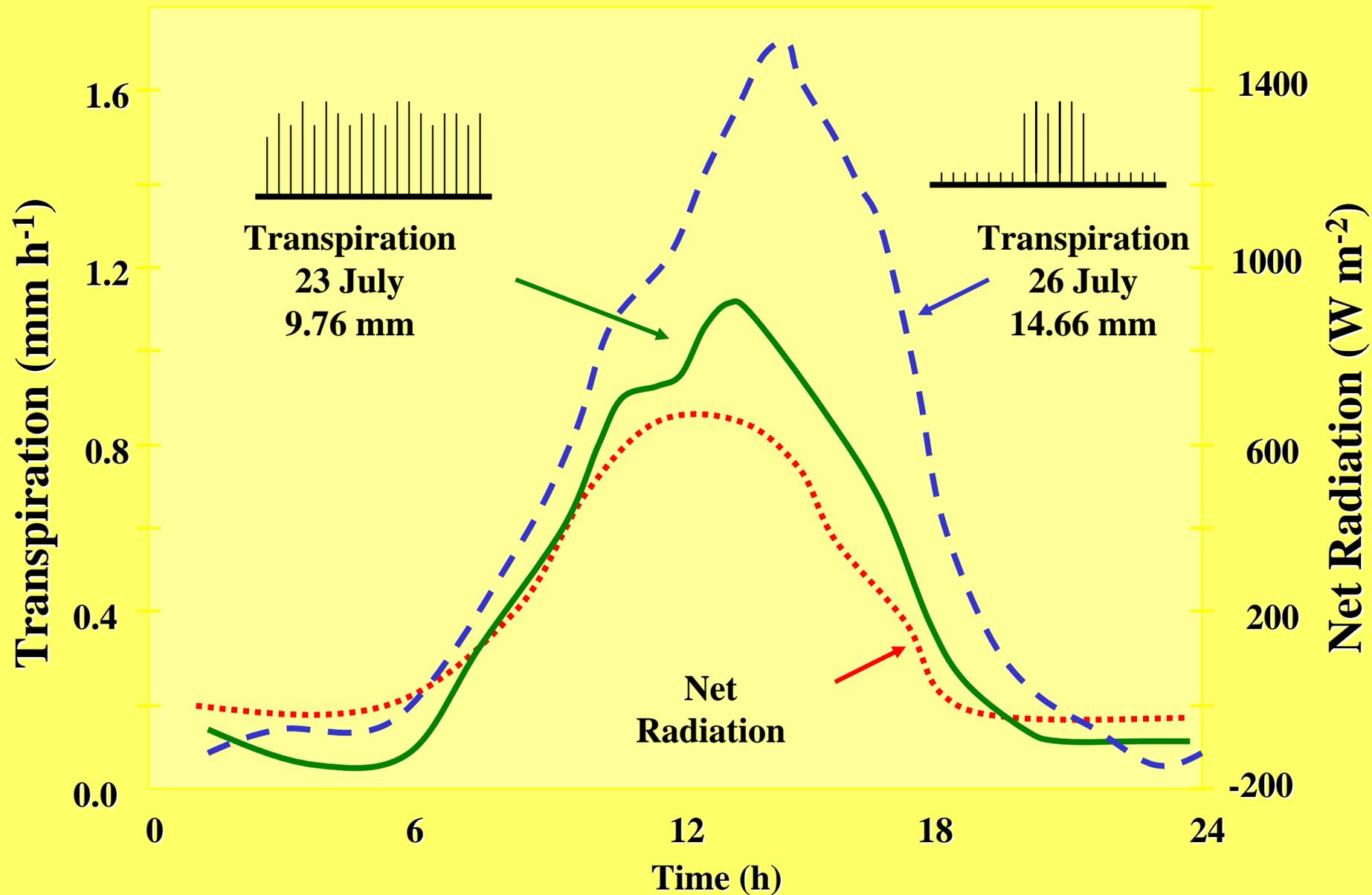


# Citrus Energy Balance - Lindsay 2001

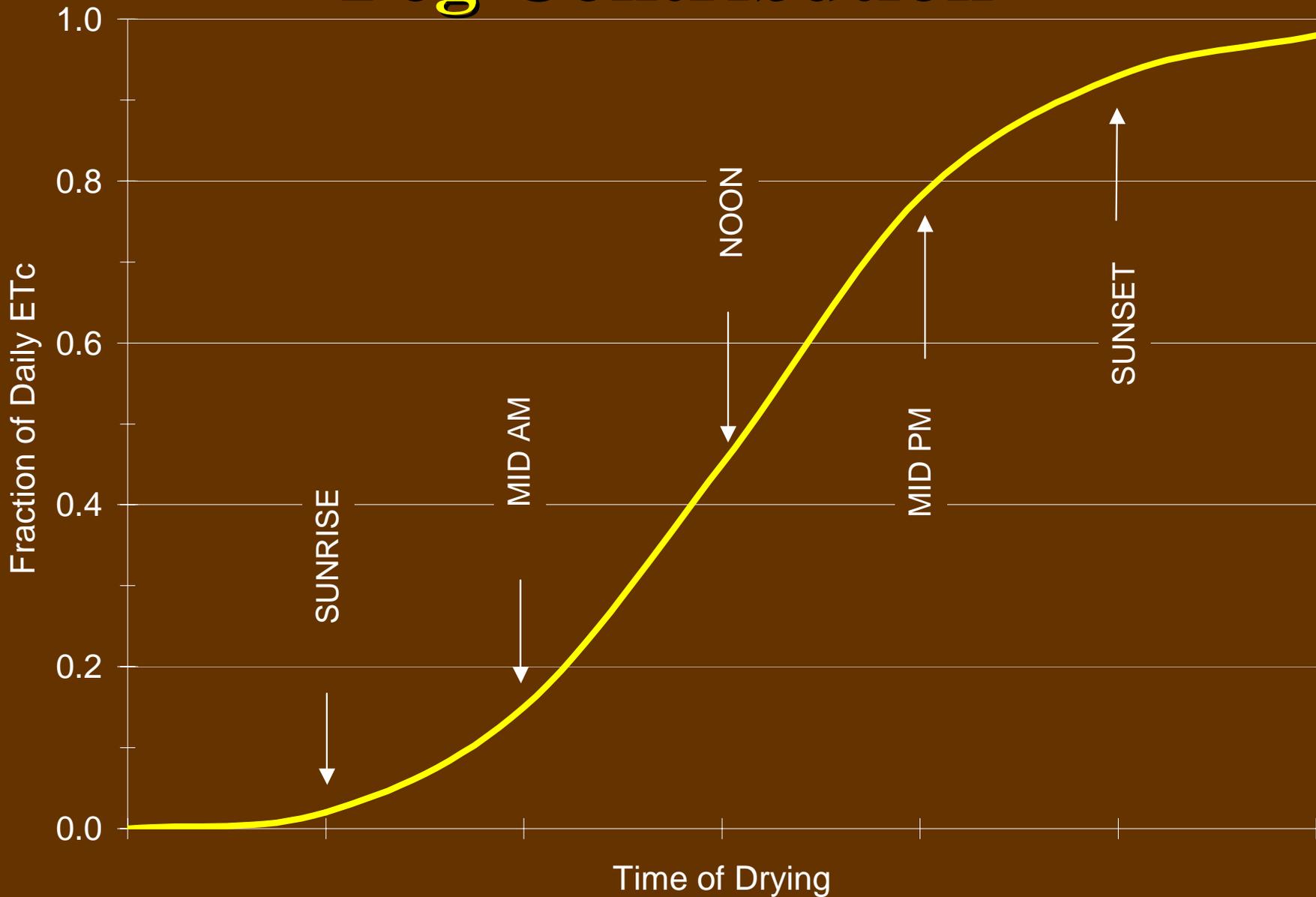
— Rn    ● LE    ◆ H    ▲ G2



# Local Advection



# Fog Contribution



# Water Table Contribution

$ET_c$



No Water Table

$ET_c$



With Water Table

**The End**

**Thanks**